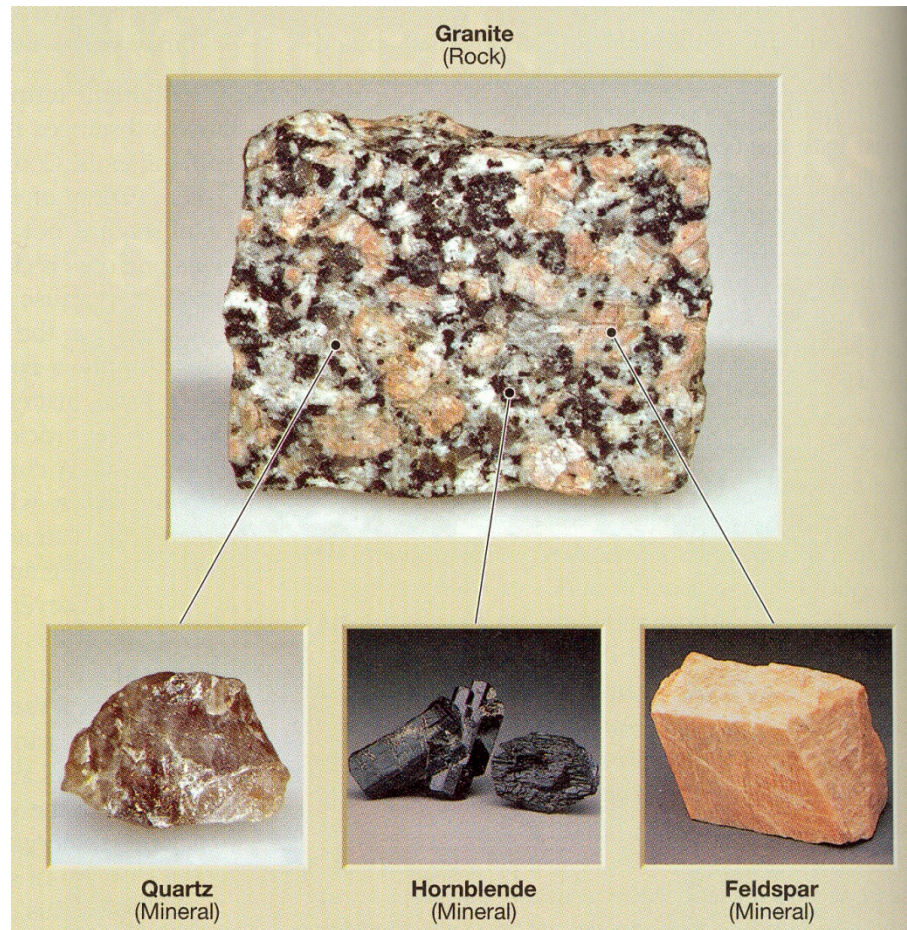


# What is a mineral?

- A mineral is a naturally occurring inorganic solid with a definite internal ordered structure.
- It should have a definite chemical composition or range of compositions.

# Minerals form rocks

- Granite made of 3 minerals

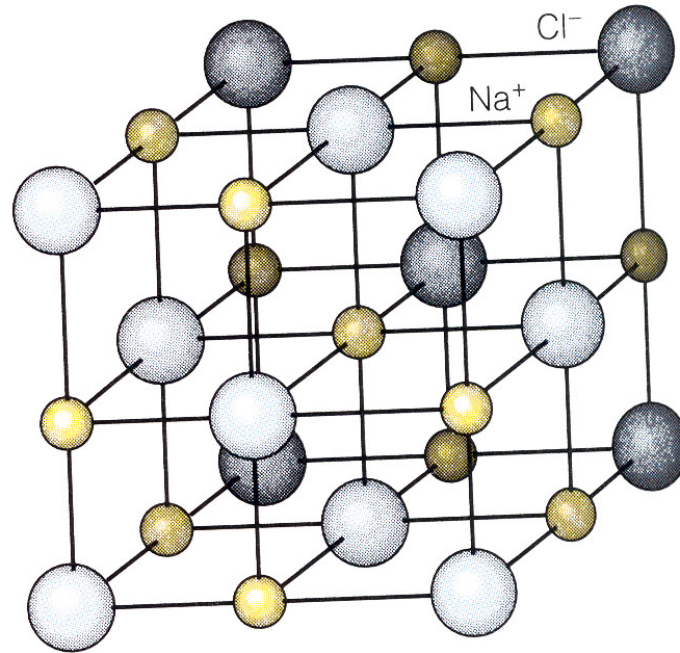
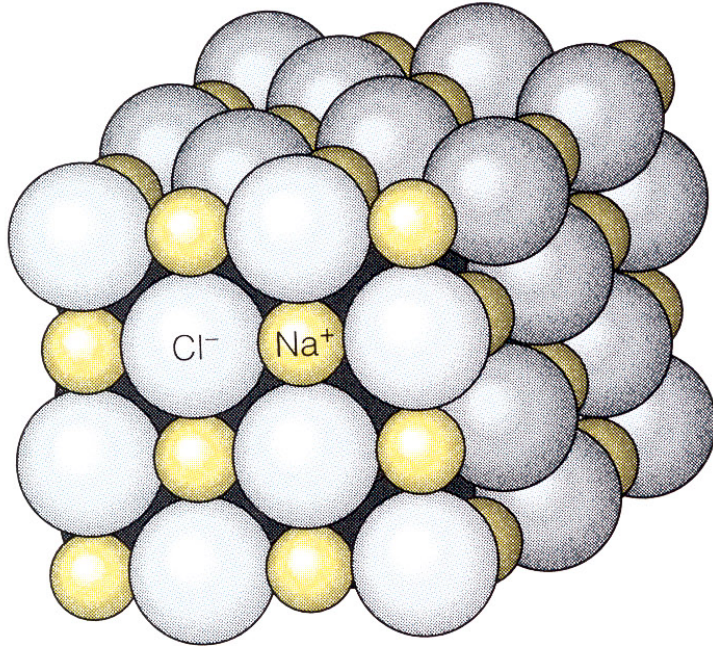


# Minerals are all around us.

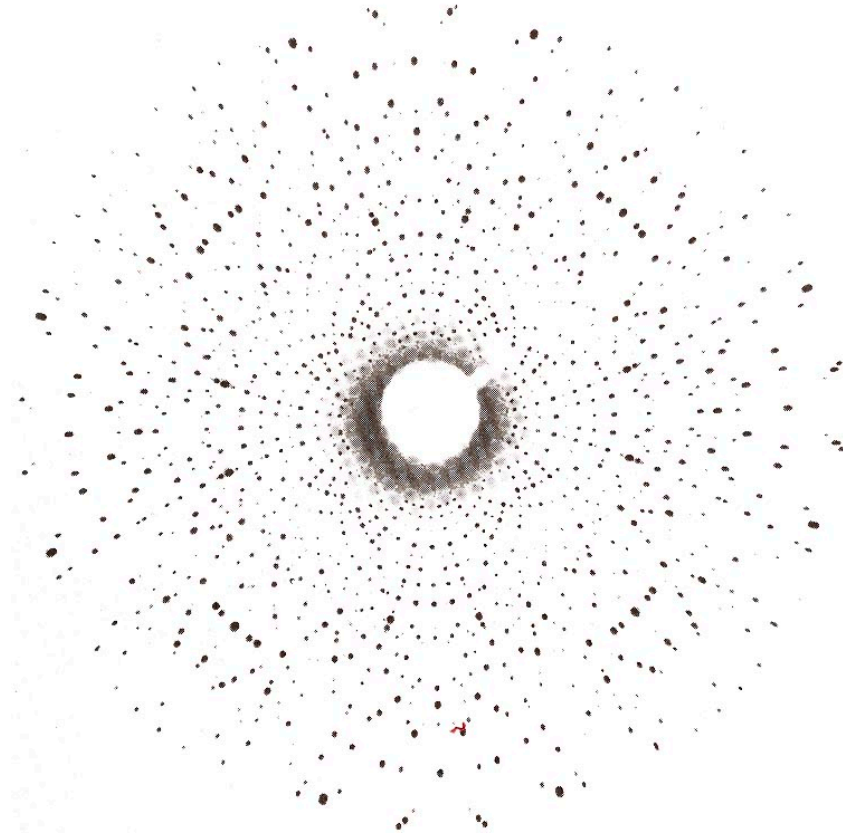
- An example of a mineral would be halite--also known as salt. It is composed of  $\text{Na}^+$  and  $\text{Cl}^-$  in equal amounts and has a cubic structure.

# Minerals

- Definite arrangement of atoms in space



# X-ray pattern of a mineral



## Definite composition

- $\text{NaCl} = \text{Halite}$
- One sodium for each Chlorine

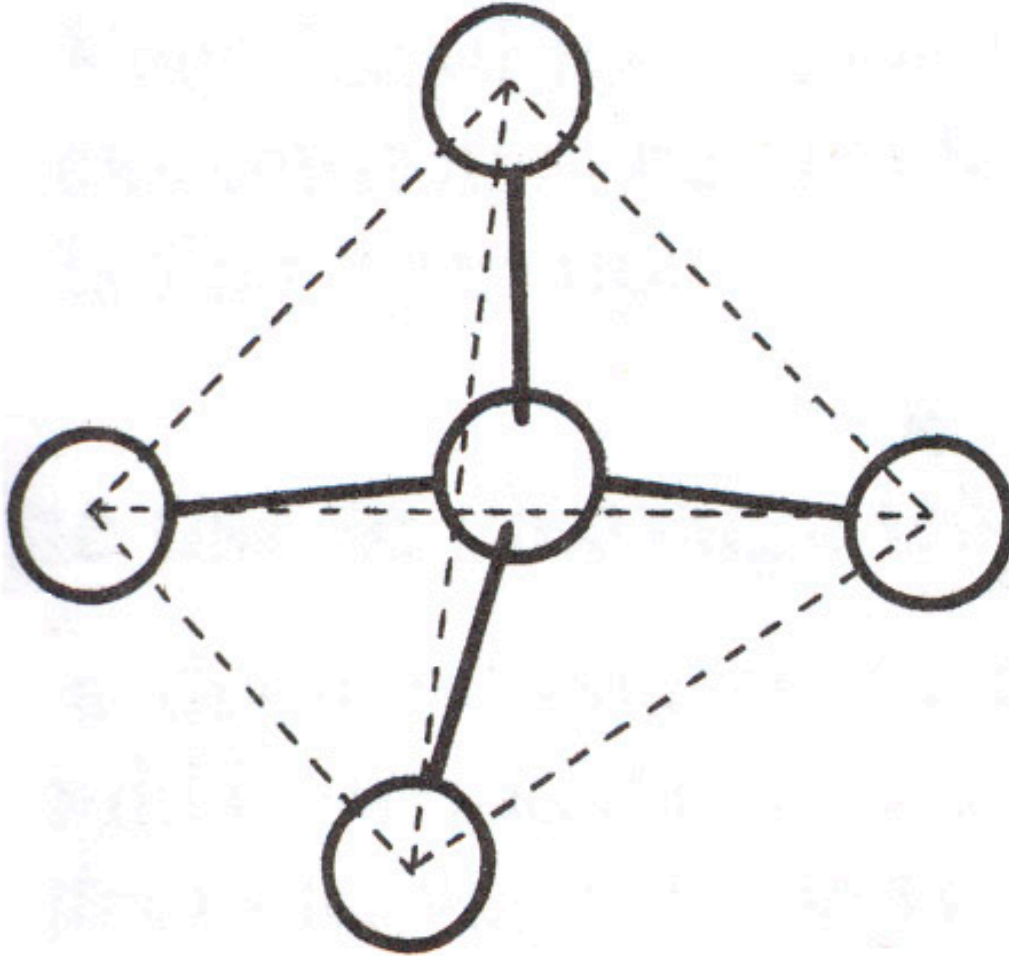


# Halite (common table salt)



# Diamond structure

tetrahedron



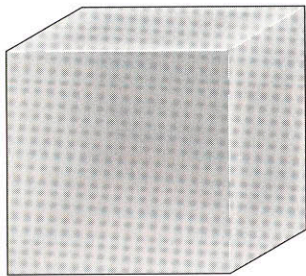


# Physical properties of minerals

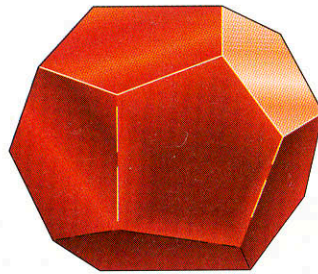
- Physical properties of minerals are a result of the minerals chemical composition and how the atoms are arranged (bonded together).
- No two minerals have the exact same physical properties, bonding, & chemical composition.

# Crystal shape

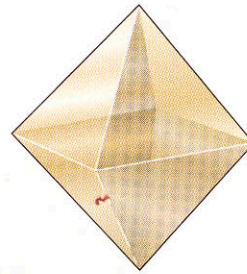
- Due to the atomic arrangement of the mineral



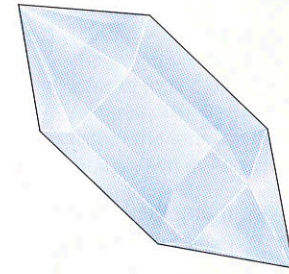
Salt



Garnet



Diamond



quartz

- As minerals grow they take on characteristic shapes

# Hardness

- Hardness is directly related to chemical bonding.
- A scale hardness called Mohs Scale of relative hardness (“Scratchability”)

1. Talc.

2. Gypsum.

3. Calcite.

4. Fluorite.

5. Apatite.

6. Orthoclase (feldspar).

7. Quartz.

8. Topaz.

9. Corundum.

10. Diamond.

# Hardness (cont...)

Friedrich Mohs based his scale on what mineral could scratch what other mineral.

**We can use a simplified scale**

- skin = 1.5 (talcum is softer)
- finger nail = 2-2.5
- penny is about 3 (calcite)
- glass = 5.5
- piece of hard steel = 6.5
- Porcelain = 6.5

# TENACITY

The resistance that a mineral offers to breaking, crushing, bending, etc.  
The following terms are used to describe tenacity in minerals:

- 1. *Brittle*. A mineral that breaks or powders easily.
- 2. *Malleable*. A mineral that can be hammered out into thin sheets.
- 3. *Sectile*. A mineral that can be cut into thin shavings with a knife.
- 4. *Ductile*. A mineral that can be drawn into wire.
- 5. *Flexible*. A mineral that bends but does not resume its original shape when the pressure is released.
- 6. *Elastic*. A mineral that, after being bent, will resume its original position upon the release of the pressure.



# Streak

- **Streak** is the color of the mineral when it is powdered using a porcelain tile. Fool's gold is black. Real gold is gold.

# Streak

- **Streak** is the color powdered mineral. A porcelain tile is used. Fool's gold is black. Real gold is gold.





# Cleavage and Fracture

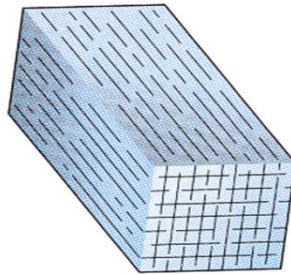
- **Cleavage** is the property that some minerals have of splitting along planes of natural weakness in the crystal structure.
- Minerals can also fracture if they have no strong preferred plane of weakness. A term Conchoidal fracture is used and you have all seen this in shards of broken glass.

# Cleavage

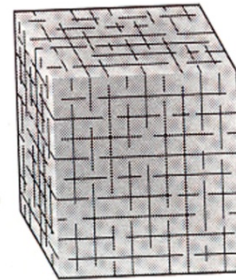
- Examples of cleavage of minerals



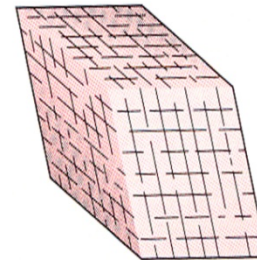
One direction



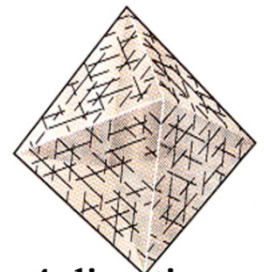
2 directions



3 directions



3 directions



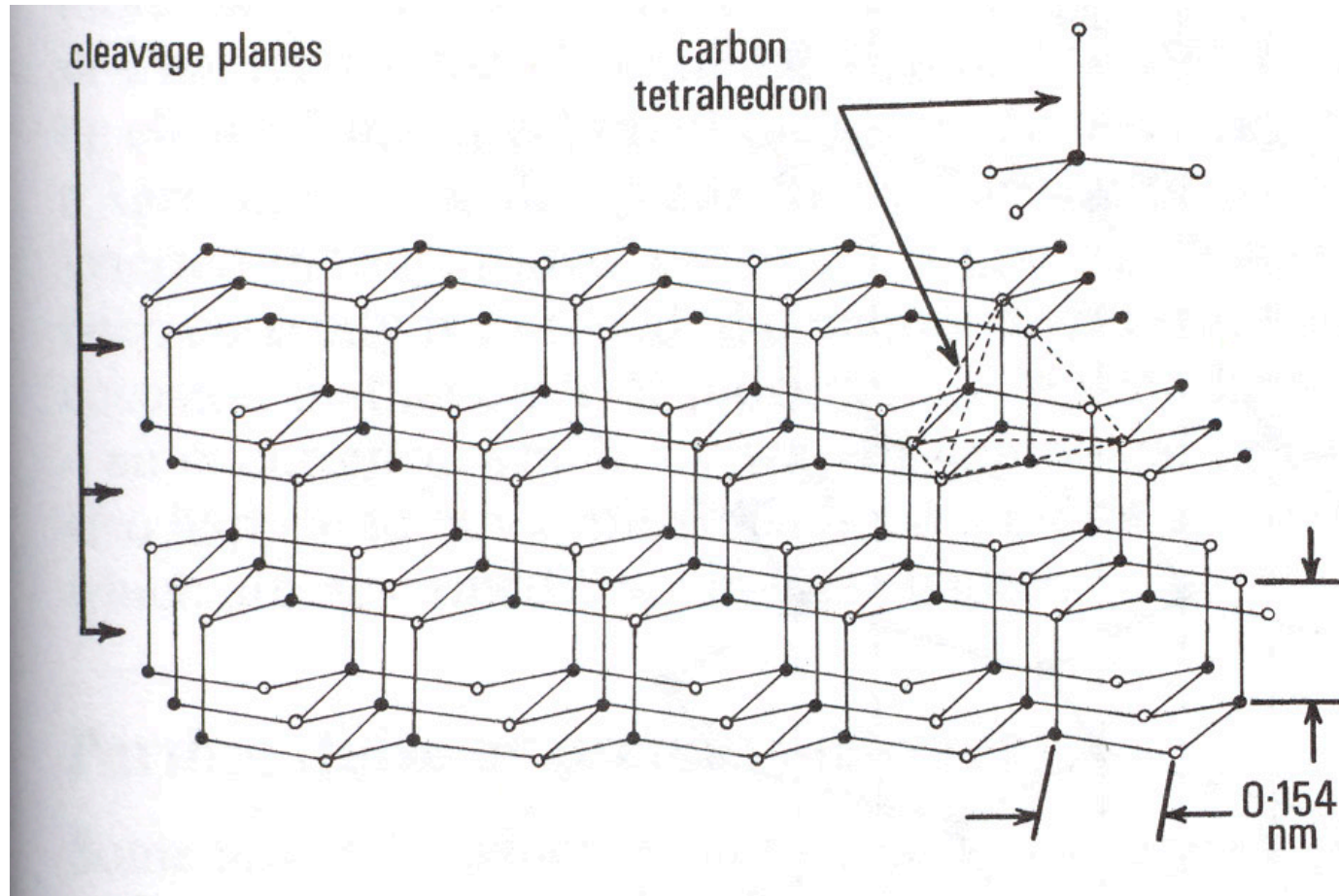
4 directions

- Mica                      feldspar                      halite                      calcite                      diamond





# Cleavage in Diamond



# Luster

- The appearance of a mineral in reflected light.
- Several specific terms are used, but it is still somewhat subjective unless the person is trained in the laboratory with examples.

# Luster (cont...)

- **Metallic**--The type of very high luster associated with metals (e.g. gold, silver, platinum) and seen in some metallic compounds (e.g. pyrites [fool's gold])
- **Adamantine**--The high surface polish achieved with diamond (zircon classified as 'sub-adamantine')
- **Vitreous**--A glass-like luster typical of the majority of gemstones (ex. quartz, sapphire, emerald, etc.)
- Resinous --The more subdued polish as seen in amber
- Waxy--The almost matte surface typical of turquoise and jadeite
- Greasy--The appearance of soapstone and nephrite (jade)
- Pearly--The luster seen in mother-of-pearl
- Silky--A fibrous luster typical of satin spar and ulexite

Table 8.2 p. 76 of Read

# Fluorescence

- How a mineral looks when viewed in the dark using ultraviolet light (UV) which can excite fluorescence
- 2 types of UV exist: short wave (SW) and long-wave (LW) UV (p. 133-135 of P.G. Read)
- SW can be dangerous to your eyes and skin!

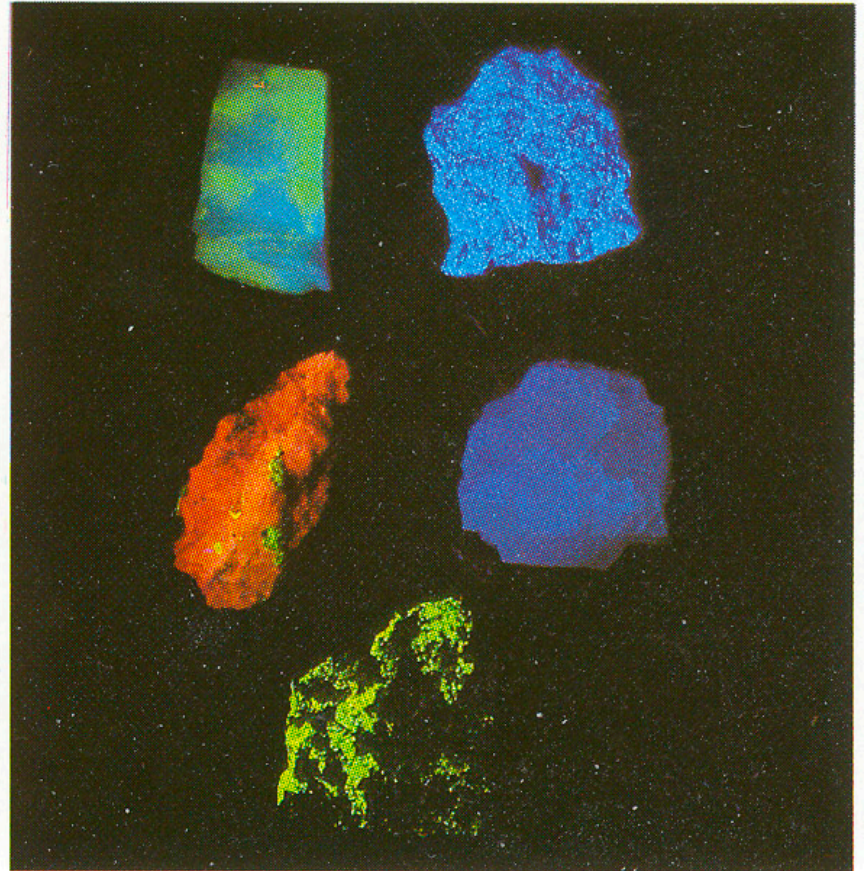


## Fluorescence (cont...)

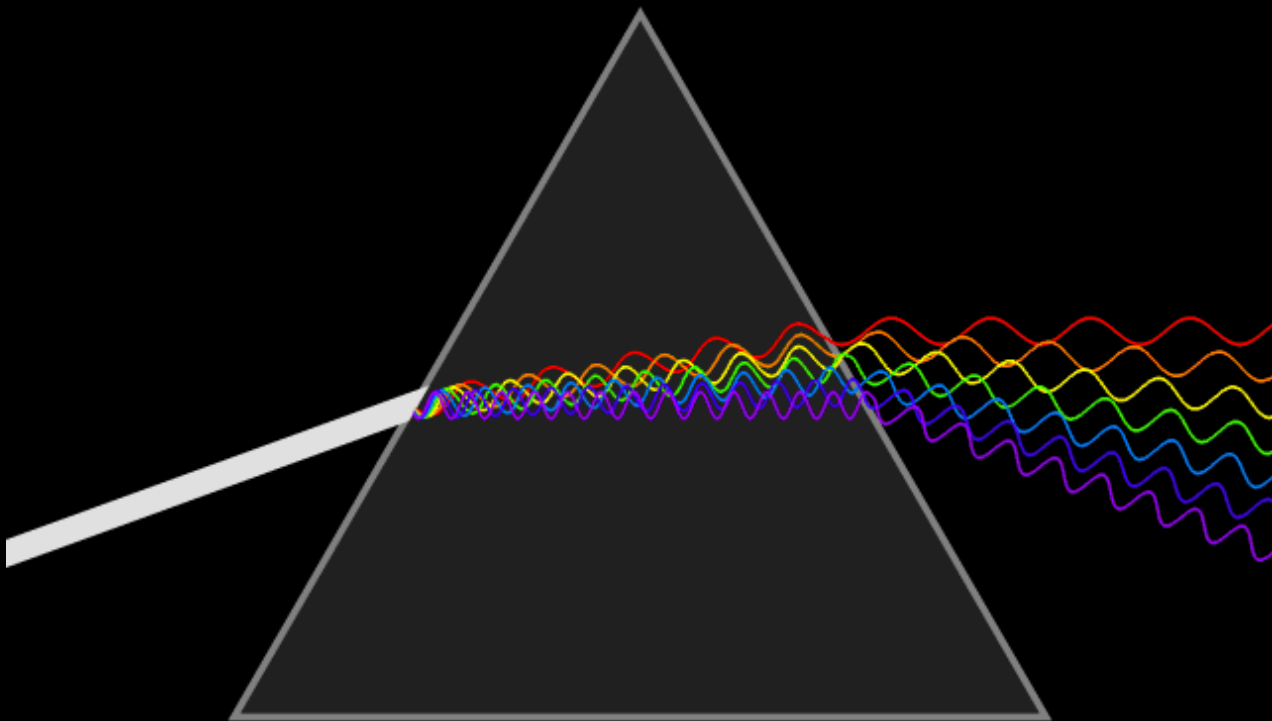
- The energy used to excite fluorescence can also stimulate a mineral to phosphoresce.
- Phosphorescence is the continued glow (emission of light) after the exciting source is turned off, for example in a luminous watch face.

# Fluorescence

aragonite, calcite; center, fluorite, halite; bottom, willemite.

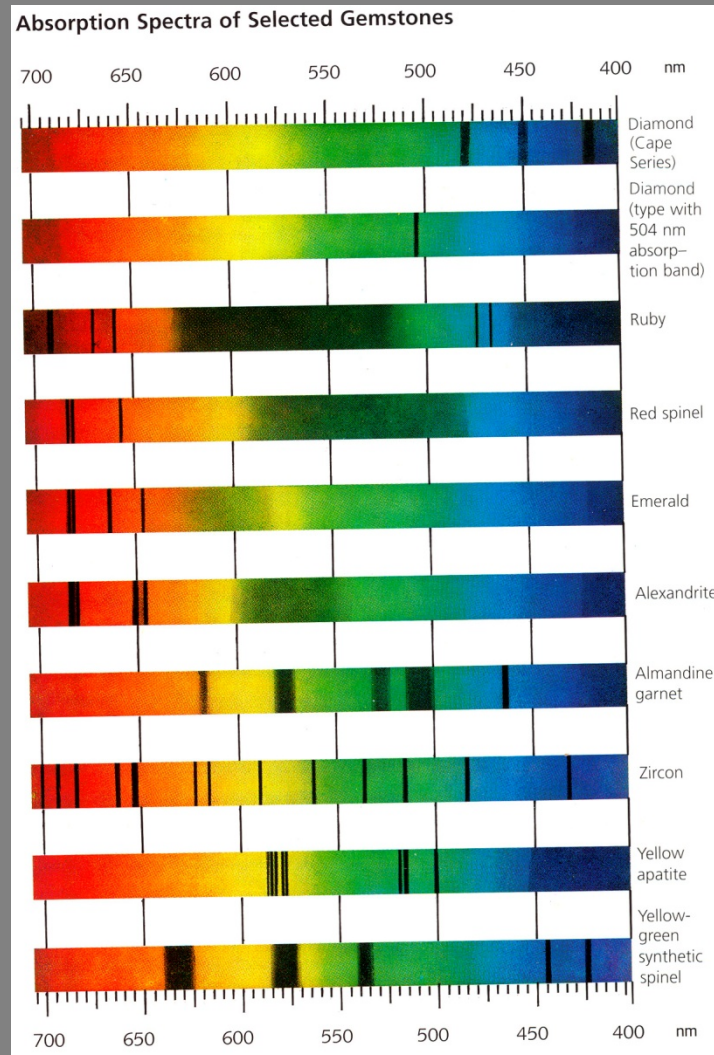


# Dispersion of light in a prism





# Spectra as seen with a spectroscope



# Specific Gravity

*Specific gravity*(G) = *density* is a number that expresses the ratio between the weight of a substance and the weight of an equal volume of water at 4°C. Thus a mineral with a specific gravity of 2, weighs twice as much as the same volume of water. The specific gravity of a mineral is frequently an important aid in its identification, particularly in working with fine crystals or gemstones, when other tests would injure the specimens.



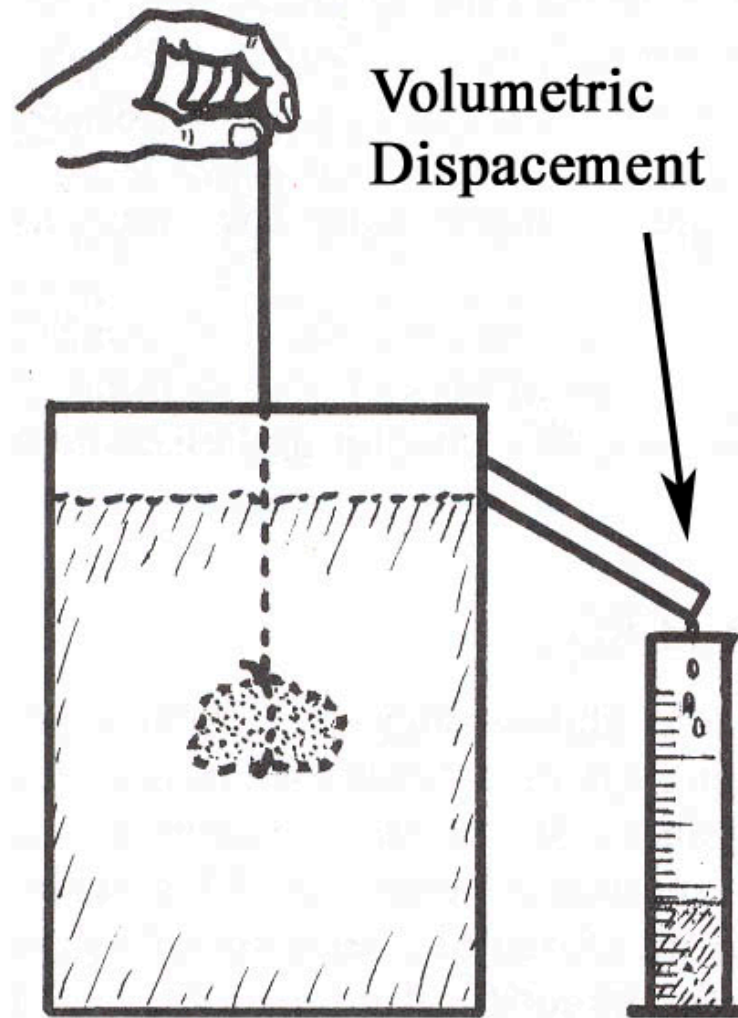
# Calculation of specific gravity

$$\text{SG of gem} = \frac{\text{weight of gem}}{\text{weight of displaced water}} = \frac{W_1}{A - B}$$

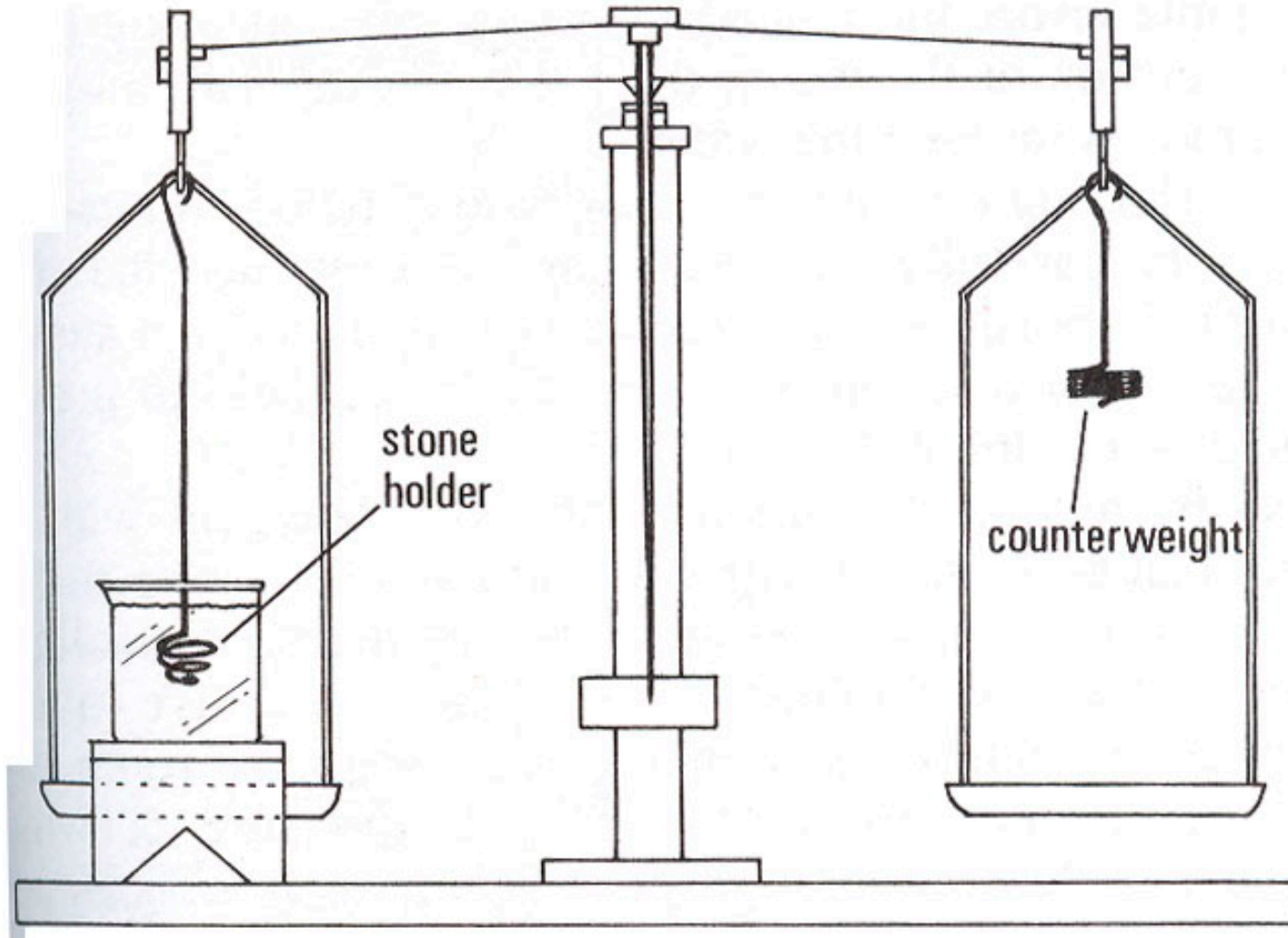
p. 58

By convention 1 gram of water has a volume of 1 milliliter (1cc) at 4°C

# Specific Gravity



# Hydrostatic method of SG Determination



## 2. Cleavage not prominent

Color	G.	H.	Remarks	Name, Composition, Crystal System
Colorless, white, smoky, Variously colored	2.65	7	Crystals usually show horizontally striated prism with rhombohedral terminations.	<b>QUARTZ</b>  $\text{SiO}_2$  Rhombohedral
Colorless, white, pale yellow	2.97 to 3.02	7	In prismatic crystals resembling topaz but distinguished by lack of good cleavage. Also in irregular masses and indistinct crystals. A rare mineral.	Danburite  $\text{Ca}(\text{B}_2\text{Si}_2\text{O}_8)$  Orthorhombic
White, colorless	2.97 to 3.0	$7\frac{1}{2}$ –8	In small rhombohedral crystals. A rare mineral.	Phenacite  $\text{Be}_2(\text{SiO}_4)$  Rhombohedral
White and almost any color	3.95 to 4.1	9	Luster adamantine to vitreous. Parting fragments may appear nearly cubic. In rude barrel-shaped crystals.	<b>CORUNDUM</b>  $\text{Al}_2\text{O}_3$  Rhombohedral
Red, black, blue, green, brown	3.6 to 4.0	8	In octahedrons; twinning common. Associated with crystalline lime- stones.	<b>SPINEL</b>  $\text{MgAl}_2\text{O}_4$  Isometric
Bluish green, yellow, pink, colorless	2.75 to 2.8	$7\frac{1}{2}$ –8	Commonly in hexagonal prisms terminated by the base; pyramid faces are rare. Crystals large in places. Poor basal cleavage.	<b>BERYL</b>  $\text{Be}_3\text{Al}_2(\text{Si}_6\text{O}_{18})$  Hexagonal
Yellowish to emerald- green	3.65 to 3.8	$8\frac{1}{2}$	In tabular crystals frequently in pseudo-hexagonal twins. Found in pegmatites.	<b>CHRYSOBERYL</b>  $\text{BeAl}_2\text{O}_4$  Orthorhombic
Green, brown, blue, red, pink, black	3.0 to 3.25	7– $7\frac{1}{2}$	In slender prismatic crystals with triangular cross section. Found usually in pegmatites. Black most common, other colors associated with lithium minerals.	<b>TOURMALINE</b>  Rhombohedral